

**SALTERS ADVANCED CHEMISTRY**  
**"CHECK YOUR NOTES" : ASPECTS OF AGRICULTURE**

Print the list below and tick the box supplied when you have covered the topic in your notes. Most of the points are covered in the **Chemical Ideas**, with supporting information in the **Storyline** or **Activities**. However, if the main source of information is the **Storyline** or an **Activity**, this is indicated.

1	The ways in which chemists can help food production, including providing extra nutrients, controlling soil pH and controlling pests (the <b>Storyline</b> in general).	<input type="checkbox"/>
2	The effect of temperature on the rate constant of the reaction.	<input type="checkbox"/>
3	The redox reactions involved in the interconversion of the following species in the nitrogen cycle: nitrogen gas, nitrate(V) ion, nitrate(III) ion, ammonium ion, dinitrogen oxide ( $\text{N}_2\text{O}$ ), nitrogen monoxide ( $\text{NO}$ ), nitrogen dioxide ( $\text{NO}_2$ ) ( <b>Storyline AA3; Activity AA3.2</b> ).	<input type="checkbox"/>
4	An outline of the manufacture of ammonia by the Haber Process, including essential conditions ( <b>Storyline AA3</b> ).	<input type="checkbox"/>
5	Why the conditions in the Haber Process are chosen, including the effect the conditions have on the position of equilibrium and on the rate of reaction ( <b>Storyline AA3</b> ).	<input type="checkbox"/>
6	The expression for the equilibrium constant, $K_p$ , for reactions involving gases (in terms of partial pressures).	<input type="checkbox"/>
7	How values of $K_p$ , together with given data on partial pressures, are used to carry out calculations concerning the composition of equilibrium mixtures.	<input type="checkbox"/>
8	The trends in reactions of the elements, and the properties of their compounds, across a period in terms of structure and bonding, including: the reactions of the elements with oxygen, chlorine and water; the acid-base character of oxides; the behaviour of chlorides towards water.	<input type="checkbox"/>
9	The relationship between the structure and bonding of a substance and its properties.	<input type="checkbox"/>
10	The partition equilibrium that occurs when a solute is distributed between two immiscible solvents.	<input type="checkbox"/>
11	The design of pesticides that combine maximum efficacy with minimum environmental damage ( <b>Storyline AA4</b> ).	<input type="checkbox"/>